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#### **Key indicators**

Single-crystal X-ray study T = 273 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.052 wR factor = 0.150 Data-to-parameter ratio = 8.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# (1*R*,4*R*)-1-(Hydroxydiphenylmethyl)-7,7-dimethylbicyclo[2.2.1]heptan-2-one

The title compound,  $C_{22}H_{24}O_2$ , has been obtained by a Grignard reaction of (1S,4R)-methyl 7,7-dimethyl-2-oxobicyclo[2.2.1]heptane-1-carboxylate with phenylmagnesium bromide. Intramolecular hydrogen bonding is observed between the carbonyl group and the hydroxy group.

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# Comment

The title compound, (I), is an important intermediate in the synthesis of *exo*-10,10-diphenyl-2,10-camphanediol, which is widely used in asymmetric reduction as a chiral auxiliary (Chen *et al.*, 1999). The molecular structure of (I) is shown in Fig. 1. Bond lengths and angles in (I) are in agreement with those reported for similar compounds (Chen *et al.*, 2006). The dihedral angle between the phenyl planes is 72.10 (6)°.



Intramolecular hydrogen bonding is observed between carbonyl groups and hydroxy groups; O2-H2A = 0.82 Å,  $H2A \cdots O1 = 2.02$  Å,  $O2 \cdots O1 = 2.722$  (3) Å and  $O2-H2A \cdots O1 = 143^{\circ}$ .

## **Experimental**

To a precooled solution of (1S,4R)-methyl 7,7-dimethyl-2-oxobicyclo[2.2.1]heptane-1-carboxylate (0.78 g, 4 mmol) in dry tetrahydrofuran (30 ml) at 273 K was added dropwise a 3 *M* solution of phenylmagnesium bromide (6.8 ml, 20 mmol) in tetrahydrofuran. The cooling bath was removed and the mixture warmed to 308 K for 10 h. The reaction was quenched with a saturated NH<sub>4</sub>Cl solution (16 ml) and extracted with ethyl acetate. The organic layer was washed with brine, dried over anhydrous MgSO<sub>4</sub>, concentrated under vacuum and the crude product was purified by column chromatography (petroleum ether–ethyl acetate, 30:1) to give the title compound as a white solid in 58% yield. Single crystals of (I) were obtained by slow evaporation of a petroleum ether–ethyl acetate solution (15:1  $\nu/\nu$ ).

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# organic papers

#### Crystal data

 $\begin{array}{l} C_{22}H_{24}O_2\\ M_r = 320.41\\ Orthorhombic, P2_12_12_1\\ a = 9.168 \ (2) \ \text{\AA}\\ b = 10.053 \ (2) \ \text{\AA}\\ c = 18.405 \ (4) \ \text{\AA}\\ V = 1696.3 \ (6) \ \text{\AA}^3 \end{array}$ 

#### Data collection

Bruker APEX area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: none 9220 measured reflections

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.052$   $wR(F^2) = 0.150$  S = 1.191912 reflections 217 parameters H-atom parameters constrained Z = 4  $D_x$  = 1.255 Mg m<sup>-3</sup> Mo K $\alpha$  radiation  $\mu$  = 0.08 mm<sup>-1</sup> T = 273 (2) K Chunk, colorless 0.65 × 0.56 × 0.41 mm

1912 independent reflections 1848 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.021$  $\theta_{\text{max}} = 26.0^{\circ}$ 

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0967P)^{2} + 0.1778P]$ where  $P = (F_{o}^{2} + 2F_{c}^{2})/3$  $(\Delta/\sigma)_{max} < 0.001$  $\Delta\rho_{max} = 0.20 \text{ e} \text{ Å}^{-3}$  $\Delta\rho_{min} = -0.18 \text{ e} \text{ Å}^{-3}$ 

Methyl H atoms were placed in calculated positions, with C–H = 0.96 Å and  $U_{iso}(H) = 1.5U_{eq}(C)$ . Other H atoms were placed in idealized positions, with C–H = 0.93 (aromatic), 0.98 (methine), 0.97 Å (methylene) and O–H = 0.82 Å, and refined in riding mode, with  $U_{iso}(H) = 1.2U_{eq}(C)$  or  $1.5U_{eq}(O)$ . In the absence of significant anomalous scattering effects, Friedel pairs were merged; the absolute configuration of (I) was assigned assuming that the absolute configuration of the starting material was retained during the synthesis.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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### Figure 1

The molecular structure of (I), with 50% probability displacement ellipsoids (arbitrary spheres for H atoms).

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### References

Bruker (2001). SAINT (Version 6.22), SMART (Version 5.625) and SADABS (Version 2.03). Bruker AXS Inc., Madison, Wisconsin, USA.

Chen, C.-J., Chu, Y.-Y., Liao, Y.-Y., Tsai, Z.-H., Wang, C.-C. & Chen, K. (1999). Tetrahedron Lett. 40, 1141–1144.

- Chen, W.-Z., Zeng, Q.-L., Fang, H., Gao, Y.-X. & Zhao, Y.-F. (2006). Acta Cryst. E62, 0869–0870.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.